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# Structural and magnetic properties of  $\text{RFe}_2\text{H}_5$ , hydrides ( $\text{R=Y}$ , Er)

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## **Abstract**

The structural and magnetic properties of RFe, hydrides  $(R=Y, Er)$  synthesized under high hydrogen pressures of 10 kbar and 373 K, have been investigated. Both saturated RFe<sub>2</sub>H<sub>r</sub> hydrides absorb about 5 H/f.u. and crystallize in the same type of orthorhombic structure (Imm2 space group), with  $a=5.437(1)$  Å,  $b=5.850(1)$  Å and  $c=8.083(1)$  Å for YFe, H<sub>e</sub> and  $a=5.424(1)$  Å,  $b=5.793(1)$  Å and  $c=8.009(1)$  Å for ErFe<sub>2</sub>H<sub>z</sub>. The large hydrogen pressure, necessary to reach 5 H/f.u. can be attributed to a partial filling of all types of available interstitial sites. YFe,  $H_5$  displays a weak ferromagnetic behavior, which is almost independent of the temperature. ErFe,  $H_5$ shows also a weak magnetization at high temperature whereas below 20 K its magnetization sharply increases due to Er magnetic ordering.  $\oslash$  2001 Elsevier Science B.V. All rights reserved.

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earth and transition metal has raised interest in both measurements. The results will be presented and discussed fundamental and application field (H storage, as a function of H content. batteries ... ). In  $\text{RFe}_2$  compounds, hydrogen absorption leads to the formation of several single-phase hydrides, which crystallize in different structures, derived from that **2. Experimental** of the parent compound [1,2]. Lowering of the crystal symmetry has been related to hydrogen ordering in prefer- The  $YFe<sub>2</sub>$  and  $ErFe<sub>2</sub>$  intermetallic compounds were ential interstitial sites [2]. In all  $\text{RFe}_2$  hydrides the changes synthesized by induction melting and characterized as in of the crystal structure accompanied by an increase of the Ref. [2]. The samples were found sin cell volume, leads to a change of the magnetic properties. homogeneous with a C15 cubic structure and cell parame-The increase of the Fe–Fe distances and the change of the ters  $a=7.36$  Å and 7.28 Å, respectively. Pieces of the interatomic bonds, due to various H surrounding explain intermetallic compounds were placed in the piston-cylinder this evolution [3]. Recently, Shashikala et al. [4] have high pressure apparatus described in [5] and outgased 16 h observed that  $ErFe<sub>2</sub>$  can absorb up to 5 H/f.u. under high at 100°C. Then the hydrogen, compressed to 100 MPa, was pressure, to form a hydride with an orthorhombic structure introduced from a pressure intensifier. Fu pressure, to form a hydride with an orthorhombic structure introduced from a pressure intensifier. Further compression  $(a=7.99 \text{ Å}, b=10.81 \text{ Å}, c=5.78 \text{ Å})$ . However no more (within the range up to 1.5 GPa) has been perfor information was given about the structural and the mag- using a hydraulic press of 100 t capacity. The temperature netic properties of this new hydride. One remaining of the high pressure apparatus has been kept constant to question is also whether larger amount of hydrogen can be  $100^{\circ}$ C by using a conventional temperature control system. stored in RFe<sub>2</sub> compounds under very high hydrogen After  $1-25$  days hydrogen exposure, the samples were pressures. quenched and kept into liquid nitrogen, until the X-ray and

**1. Introduction** In this purpose, YFe<sub>2</sub> and ErFe<sub>2</sub> hydrides have been submitted to hydrogen pressure of up to 10 kbar. The Hydrogen absorption in intermetallic compounds of rare resulting hydrides were analyzed by X-ray and magnetic

Ref. [2]. The samples were found single-phase and magnetic measurements could be performed. The X-ray \*Corresponding author. Fax: <sup>1</sup>33-149-781-203. diffraction (XRD) patterns were registered with a D8 *E-mail address:* paulbon@glvt-cnrs.fr (V. Paul-Boncour). Brucker diffractometer between 10 and 120° in 20 and a

step of 0.02° (Cu K<sub> $\alpha$ </sub> wavelength). They were refined with 30.6%), the *a* and *b* parameters are divided by about  $\sqrt{2}$  the Rietveld method, using the Fullprof program [6]. The whereas the *c* parameter is almost un magnetic measurements were performed from 4.2 to 290 K and H site positions were derived from that of the Fd  $\overline{3}$  m with a DSM8 Manics Magneto-susceptometer for applied space group with appropriate transformations, and could be fields up to 1.8 T. described in the space group Imm2 (No. 44) as given in

after hydrogenation at 10 kbar and 473 K, looks like the tances calculated in these orthorhombic structures are XRD pattern of ErFe<sub>2</sub>H<sub>48</sub> published by Shashikala et al. reported in Table 4. The 2 Fe sites display more different [4]. After a week, additional reflections corresponding to a surrounding than the two Y or Er atoms. This difference rhombohedral phase were observed for both hydrides, may result from a stronger influence of the interstitial indicating the occurrence of hydrogen desorption. The hydrogen on their Fe neighbors. analysis by thermal decomposition of the hydrogen content in both saturated samples, indicated about 5 H/f.u. In the 3.2. *Magnetic properties* further part of this paper the saturated hydrides will be denoted as  $YFe<sub>2</sub>H<sub>5</sub>$  and  $ErFe<sub>2</sub>H<sub>5</sub>$ . The XRD patterns of The magnetization curves of  $YFe<sub>2</sub>H<sub>5</sub>$  show a ferromag-<br> $YFe<sub>2</sub>H<sub>5</sub>$  and  $ErFe<sub>2</sub>H<sub>5</sub>$  were indexed in a centered ortho-<br>netic  $YFe<sub>2</sub>H<sub>5</sub>$  and ErFe<sub>2</sub>H<sub>5</sub> were indexed in a centered ortho-<br>rhombic cell with  $a=5.437(1)$  Å,  $b=5.850(1)$  Å and<br> $c=8.083(1)$  Å for  $YFe<sub>2</sub>H<sub>5</sub>$  and  $a=5.424(1)$  Å,  $b=$ <br>temperature dependant magnetization  $c = 8.083(1)$  Å for YFe<sub>2</sub>H<sub>5</sub> and  $a = 5.424(1)$  Å,  $b = 5.793(1)$  Å and  $c = 8.009(1)$  Å for ErFe<sub>2</sub>H<sub>5</sub>. The cell 5.793(1) Å and  $c = 8.009(1)$  Å for ErFe<sub>2</sub>H<sub>5</sub>. The cell pared to that of YFe<sub>2</sub>H<sub>4.2</sub> and YFe<sub>2</sub>H<sub>3.5</sub>. The magneti-<br>proposed in Ref. [4] could not be used to index the zation of YFe<sub>2</sub>H<sub>5</sub> is significantly reduced compa pared to the cubic cell of the  $RFe<sub>2</sub>$  parent compounds and taking into account the large cell volume increase (29 and

whereas the  $c$  parameter is almost unchanged. The R, Fe Table 1. The hydride XRD patterns were refined with this new structure and good agreement was obtained as reported for YFe<sub>2</sub>H<sub>5</sub> in Table 2 and in Fig. 1. For ErFe<sub>2</sub>H<sub>r</sub> a **3. Results** small contribution of a second cubic phase (8%) with  $a=7.90$  Å, must be added to refine correctly its XRD 3.1. *Crystal structure determination* pattern (Table 3). A hydrogen content of 5 H/f.u. leads to 3.9 Å<sup>3</sup>/H, which is in good agreement with The XRD patterns of YFe, H<sub>r</sub> and ErFe, H<sub>r</sub> compounds the values obtained for other hydrides. Interatomic dis-

zation of  $YFe<sub>2</sub>H<sub>5</sub>$  is significantly reduced compared to that  $YFe<sub>2</sub>H<sub>5</sub>$  pattern even with larger cell parameters. Com-<br>pared to the cubic cell of the RFe, parent compounds and slightly as the temperature decreases. After a week a wide transition is observed between 100 and 150 K, which can

Table 1

Change of the atomic site symmetry and coordinates of Y, Fe and H from the cubic space group Fd  $\overline{3}$  m (Non-centrosymmetric setting) to the orthorhombic space group Imm2

$S.G.$	Fd $\bar{3}$ m No. 227				Imm2 No. 44			
	Site	$\boldsymbol{x}$	$\mathcal{Y}% _{0}$	$\ensuremath{\mathnormal{z}}$	Site	$\boldsymbol{x}$	$\mathcal{Y}$	Z.
${\bf R}$	$\rm 8a$	$\mathbf{0}$	$\mathbf{0}$	$\boldsymbol{0}$	2a	$\overline{0}$	$\overline{0}$	$\boldsymbol{0}$
					2 <sub>b</sub>	$\mathbf{0}$	1/2	3/4
Fe	16d	5/8	5/8	$5/8$	4c	1/4	$\mathbf{0}$	$5/8$
					4d	$\overline{0}$	1/4	3/8
D(A2B2)	96g	$7/16$	7/16	1/4	4c	$1/\sqrt{8}$	$\mathbf{0}$	1/4
					4c	$3/8$	$\mathbf{0}$	$\boldsymbol{0}$
					4d	$\boldsymbol{0}$	1/8	3/4
					4d	$\mathbf{0}$	3/8	$\boldsymbol{0}$
					8e	3/16	5/16	15/16
					8e	5/16	5/16	5/16
					8e	3/16	5/16	9/16
					8e	3/16	3/16	3/16
D(AB3)	$32\mathrm{e}$	13/32	13/32	13/32	4c	3/16	$\mathbf{0}$	13/32
					4c	5/16	$\boldsymbol{0}$	27/32
					4d	$\overline{0}$	$3/16$	19/32
					4d	$\boldsymbol{0}$	5/16	$5/32$
D(B4)	$8\mathrm{b}$	$1/2\,$	1/2	1/2	2a	$\boldsymbol{0}$	$\mathbf{0}$	1/2
					2 <sub>b</sub>	$\mathbf{0}$	1/2	1/4







Fig. 1. Refined XRD pattern of YFe<sub>2</sub>H<sub>5</sub>, with the values of Table 2. (O) Experimental curve, (-) calculated curve, difference curve (bottom), (|) hkl line positions ( $\lambda$ =1.54178 Å). The pattern was measured with a D8 Brucker diffractometer.

be related to a small amount of  $YFe<sub>2</sub>H<sub>4.2</sub>$  due to H temperature dependent magnetization M(T) of  $ErFe<sub>2</sub>H<sub>5</sub>$ , desorption as observed in the XRD patterns. The mag- measured with different applied magnetic netization curves versus field of ErFe<sub>2</sub>H<sub>5</sub>, reported in Fig. M<sub>S</sub> magnetization extrapolated at H=0 from the M(H) 4 are not saturated. The superposed susceptibility is curves are reported on Fig. 5. The values of M<sub>S</sub> maximum at 10 K  $(1.5 \t10^{-1} \t\mu B/kOe)$  and decreases  $M(T)$  measured at 0.2 T. The sharp increase of  $M(T)$ <br>continuously down to 1.7 10<sup>-3</sup>  $\mu B/kOe$  at 290 K. The observed below 30 K, should be attributed to erbium

measured with different applied magnetic fields, and the

Table 3

Refined parameters and atomic positions for the orthorhombic and cubic phases in ErFe, $H_5$  hydride

Orthorhombic phase		S.G. Imm2		
Cell parameters	$a = 5.424(1)$ Å	$b = 5.793(1)$ Å	$c = 8.009(1)$ Å	$V = 252 \text{ Å}^3$
Atom	$\boldsymbol{x}$		Z.	$B(\AA^2)$
Er1	0	$\Omega$	$-0.012(1)$	1.3
Er2		0.5	0.753(1)	1.3
Fe1	0.241(1)	$\Omega$	0.628(1)	0.2
Fe <sub>2</sub>	$\Omega$	0.235(1)	0.368(1)	0.2
$R_{F}$ : 7.7%				
Cubic phase				
Cell parameter	$a = 7.900$ Å			$V=493 \text{ Å}$
Atom	$\mathcal{X}$	$\mathcal{V}$	Z.	$B(\AA^2)$
Er	0.125	0.125	0.125	1.4
Fe	0.5	0.5	0.5	2.8
$R_F = 8\%$				
Rp: 9.6%				

Table 4

Number of neighbors N and averaged interatomic distances between R and Fe nearest neighbors in  $\text{RFe}_2\text{H}_5$  hydrides. The values in bracket correspond to the maximum distance deviation from the average value

Central atom	Neighbor atom	N	$d(\AA)$ $R = Y$	d(A) $R = Er$
Fe1	Fe	6	2.80(0.08)	2.78(0.17)
Fe1	R	6	3.35(0.13)	3.33(0.16)
Fe <sub>2</sub>	Fe	6	2.87(0.17)	2.85(0.22)
Fe <sub>2</sub>	R	6	3.30(0.20)	3.27(0.17)
R1	Fe	12	3.32(0.18)	3.31(0.14)
R1	R	$\overline{4}$	3.48(0.01)	3.45(0.01)
R <sub>2</sub>	Fe	12	3.33(0.06)	3.29(0.15)
R <sub>2</sub>	R	4	3.48(0.01)	3.45(0.01)







Fig. 4. Magnetization versus field of  $YFe<sub>2</sub>H<sub>5</sub>$  between 4.2 and 290 K.

### **4. Discussion**

Hydrogen absorption under high pressure leads to similar orthorhombic phases for both  $YFe<sub>2</sub>H<sub>5</sub>$  and  $E$ rFe<sub>2</sub>H<sub>5</sub>. The structural distortions of the RFe<sub>2</sub> cubic cell, upon hydrogen absorption have been attributed to various filling of hydrogen in non-equivalent interstitial sites. In the cubic Fd  $\overline{3}$  m space group three different interstitial sites are available: A2B2 (96g), AB3 (32e) and B4 (8b)  $(A=R$  and  $B=Fe$ ). According to both geometrical and thermodynamical criteria the 96g sites (A2B2) sites are Fig. 2. Magnetization versus field of YFe,  $H_s$  at 6 and 290 K. preferentially occupied for low H content. The rhombohedral distortion in ErFe<sub>2</sub>H<sub>3.5</sub> (R  $\bar{3}$  m space group) was attributed to the filling of  $AB_3$  sites [1] whereas in magnetic ordering, since as observed for  $YFe<sub>2</sub>H<sub>5</sub>$  the Fe<br>magnetization should remain almost constant at low tem-<br>perature. The minimum at 20 K, observed in M(T) at 0.2 T<br>and M<sub>s</sub>(T), can be related to a compensa



Fig. 3. Temperature dependant magnetization of  $YFe<sub>2</sub>H<sub>x</sub>$  compounds for Fig. 5. Temperature dependant magnetization at different applied fields and saturation magnetization M, in ErFe, H<sub>e</sub>. and saturation magnetization  $M_s$  in ErFe<sub>2</sub>H<sub>5</sub>.

distance between two hydrogen atoms  $R_{H-H} \approx 2.1$  Å, **5. Conclusion** shows that the hydrogen content of 5 H/f.u can be reached

In YFe<sub>2</sub>H<sub>r</sub> hydrides, the magnetization is dominated by the magnetization observed for both compounds compared to Fe–Fe interaction and  $M_{Fe}$  is reduced from 1.95  $\mu$ B for hydrides with lower H content, may result either from a  $x=3.6$  to 0.4  $\mu$ <sub>R</sub> for  $x=5$ . This decrease of the bulk fanning of the Fe and Er moments, or from a n  $x=3.6$  to 0.4  $\mu_B$  for  $x=5$ . This decrease of the bulk fanning of the Fe and Er moments, or from a non-collinear magnetization can result either from a reduction of the Fe orientation of the magnetic moments. In additi magnetization can result either from a reduction of the Fe moment or to a non-collinear orientation of the Fe mo-<br>ments as proposed by Shashikala et al. [8]. Since in the indicates a lowering of the exchange interaction between orthorhombic structure, there are two different Fe sites Er and Fe moments. with different surrounding, two different orientations of the Fe moments cannot be excluded. In  $YFe<sub>2</sub>D<sub>4.2</sub>$  the metamagnetic transition observed at 90 K [3] can result **References** from a change of Fe moment orientation under an external field. In  $E$ rFe<sub>2</sub>H<sub>5</sub>, the small value of the compensation [1] D. Fruchart, Y. Berthier, T. de Saxce, P. Vuillet, J. Alloys Comp. 130 temperature T (20 K) observed for  $E$ rFe<sub>2</sub>H<sub>z</sub> agrees (1987) 89. temperature  $T_{\text{comp.}}$  (20 K) observed for ErFe<sub>2</sub>H<sub>5</sub> agrees (1987) 89.<br>
well with the reduction of  $T_{\text{comp.}}$  versus H content ob-<br>
served in ErFe<sub>2</sub>H<sub>x</sub> compounds by Fruchart et al. [1]. This (20 V. Paul-Boncour, L. Gu decrease reflects the lowering of the exchange interaction [3] V. Paul-Boncour, A. Percheron-Guégan, J. Alloys Comp. 293–295 between rare earth and transition metal as the interatomic (1999) 237. distances increase. A fanning of the Er moment has been [4] K. Shashikala, P. Raj, A. Sathyamoorthy, Mater. Res. Bull. 31<br>character is the studied at al. [0] in E-Ea H asseming an (1996) 957. observed by Fish et al. [9] in  $Erfe_2H_{3.5}$  assuming an<br>antiparallell collinear coupling of the Er and Fe moments [6] J. Rodriguez-Carjaval (Ed.), in: Union of Crystallography, Abstracts as in ErFe<sub>2</sub>. However since there are two different Er sites of Satellite Meeting on Powder Diffraction, Toulouse, France, 1990, in orthorhombic  $Erfe_2H_5$ , a non-collinear orientation of p. 127.<br>the Er moments can be expected. The large value of the [7] M. Latroche, V. Paul-Boncour, A. Percheron-Guégan, F. Bouréethe Er moments can be expected. The large value of the  $[7]$  M. Latroche, V. Paul-Boncour, A. Percheron-G<br>
Vigneron, J. Solid State Chem. 133 (1997) 568. superposed susceptibility in the M(H) curves may support<br>this assumption. Further experiments, like  $^{57}$ Fe Mössbauer [8] J.J. Rhyne, G.E. Fish, S.G. Sankar, W.E. Wallace, J. Physique 40<br>(1979) C5–209. spectroscopy and neutron diffraction experiments will be [9] K. Shashikala, P. Raj, A. Sathyamoorthy, T.V. Chandradekhar Rao, V. performed to check the origin of the strong reduction of Siruguri, S.K. Paranjpe, Phil. Mag. B 79 (1999) 1195. the bulk magnetization observed in both  $YFe<sub>2</sub>H<sub>5</sub>$  and  $E$ rFe<sub>2</sub>H<sub>5</sub> hydrides, compared to the hydrides with lower H content.

only if the three types of A2B2, AB3 and B4 sites are The synthesis of  $RFe<sub>2</sub>$  hydrides ( $R = Y$ , Er) under 10 partially occupied. With only A2B2 sites we reach about 3 kbar hydrogen, leads to the formation of orthorhombic H/f.u. and with A2B2 and AB3 sites 4 H/f.u. as maximal RFe,  $H_5$  hydrides. Their crystal structures were refined in H content. Since the B4 sites are thermodynamically less the space group Imm2 with two different R and Fe sites. stable than the A2B2 and AB3 sites, a very high hydrogen The analysis of the possible interstitial hydrogen sites pressure is therefore required to fill these new sites. shows that according to geometrical criterion all type of Bulk magnetization of  $YFe<sub>2</sub>H<sub>5</sub>$  and ErFe<sub>2</sub>H<sub>5</sub> are strong-<br>1. A2B2, AB3 and B4 interstitial sites should be occupied to 25 1/ 1. The reduction of the bulk reach the concentration 5 H/f.u. The reduction of the bulk indicates a lowering of the exchange interaction between

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